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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
08/634,255	04/18/1996	NORIO OHKUMA	35.C11365	9044

5514 7590 04/02/2002

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NEW YORK, NY 10112

EXAMINER

BROOKE, MICHAEL S

ART UNIT	PAPER NUMBER
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2853

DATE MAILED: 04/02/2002

Please find below and/or attached an Office communication concerning this application or proceeding.

Office Action Summary

Application N .

08/634,255

Applicant(s)

OHKUMA ET AL.

Examiner

Michael S. Brooke

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-- The MAILING DATE of this communication appears on the cover sheet with the correspondence address --

Period for Reply

A SHORTENED STATUTORY PERIOD FOR REPLY IS SET TO EXPIRE 3 MONTH(S) FROM THE MAILING DATE OF THIS COMMUNICATION.

- Extensions of time may be available under the provisions of 37 CFR 1.136(a). In no event, however, may a reply be timely filed after SIX (6) MONTHS from the mailing date of this communication.
- If the period for reply specified above is less than thirty (30) days, a reply within the statutory minimum of thirty (30) days will be considered timely.
- If NO period for reply is specified above, the maximum statutory period will apply and will expire SIX (6) MONTHS from the mailing date of this communication.
- Failure to reply within the set or extended period for reply will, by statute, cause the application to become ABANDONED (35 U.S.C. § 133).
- Any reply received by the Office later than three months after the mailing date of this communication, even if timely filed, may reduce any earned patent term adjustment. See 37 CFR 1.704(b).

Status

- 1) ☒ Responsive to communication(s) filed on 04 March 2002.
- 2a) ☐ This action is FINAL. 2b) ☒ This action is non-final.
- 3) ☐ Since this application is in condition for allowance except for formal matters, prosecution as to the merits is closed in accordance with the practice under *Ex parte Quayle*, 1935 C.D. 11, 453 O.G. 213.

Disposition of Claims

- 4) ☒ Claim(s) 1,2 and 4-15 is/are pending in the application.
- 4a) Of the above claim(s) _____ is/are withdrawn from consideration.
- 5) ☐ Claim(s) _____ is/are allowed.
- 6) ☒ Claim(s) 1,2 and 4-15 is/are rejected.
- 7) ☐ Claim(s) _____ is/are objected to.
- 8) ☐ Claim(s) _____ are subject to restriction and/or election requirement.

Application Papers

- 9) ☐ The specification is objected to by the Examiner.
- 10) ☐ The drawing(s) filed on _____ is/are: a) ☐ accepted or b) ☐ objected to by the Examiner.
- Applicant may not request that any objection to the drawing(s) be held in abeyance. See 37 CFR 1.85(a).
- 11) ☐ The proposed drawing correction filed on _____ is: a) ☐ approved b) ☐ disapproved by the Examiner.
- If approved, corrected drawings are required in reply to this Office action.
- 12) ☐ The oath or declaration is objected to by the Examiner.

Priority under 35 U.S.C. §§ 119 and 120

- 13) ☐ Acknowledgment is made of a claim for foreign priority under 35 U.S.C. § 119(a)-(d) or (f).
- a) ☐ All b) ☐ Some * c) ☐ None of:
1. ☐ Certified copies of the priority documents have been received.
2. ☐ Certified copies of the priority documents have been received in Application No. _____.
3. ☐ Copies of the certified copies of the priority documents have been received in this National Stage application from the International Bureau (PCT Rule 17.2(a)).
- * See the attached detailed Office action for a list of the certified copies not received.
- 14) ☐ Acknowledgment is made of a claim for domestic priority under 35 U.S.C. § 119(e) (to a provisional application).
- a) ☐ The translation of the foreign language provisional application has been received.
- 15) ☐ Acknowledgment is made of a claim for domestic priority under 35 U.S.C. §§ 120 and/or 121.

Attachment(s)

- 1) ☐ Notice of References Cited (PTO-892)
- 2) ☐ Notice of Draftsperson's Patent Drawing Review (PTO-948)
- 3) ☐ Information Disclosure Statement(s) (PTO-1449) Paper No(s) _____
- 4) ☐ Interview Summary (PTO-413) Paper No(s). _____
- 5) ☐ Notice of Informal Patent Application (PTO-152)
- 6) ☐ Other:

DETAILED ACTION

Claim Rejections - 35 USC § 103

1. The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:

(a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negated by the manner in which the invention was made.

2. Claims 1, 2 and 4-15 are rejected under 35 U.S.C. 103(a) as being unpatentable over Ohkuma et al. (U.S. 5,478,606) in view of Nakahata et al. (5,166,265).

Ohkuma et al. discloses a liquid jet recording head which includes a member formed from a cured product of a resin composition comprising an epoxy and a photopolymerization initiator which acts to cure the epoxy (see column 5, lines 35-60). The epoxy compound is an aromatic epoxy compound such as bisphenol A (see column 5, lines 35-36). The curable epoxy compound disclosed also includes an alicyclic epoxy having an oxycyclohexane skeleton (see column 5, lines 35-42). The reference also discloses a method of making the liquid jet recording head which entails forming an ink flow path pattern from a soluble resin on an ink discharge pressure-generating element on a base plate, forming a coating resin layer on the soluble resin layer, removing of the soluble resin layer by elution, and forming a discharge opening through the coating resin layer (see column 2, lines 28-42). In addition, the reference discloses that the method of forming the discharge opening is accomplished by the well known technique of photolithography (see column 4, lines 28-32). Finally, Ohkuma et al. discloses in

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column 7, lines 29-30 that the discharge openings can be formed by either oxygen plasma etching or excimer laser etching.

Ohkuma et al. discloses the claimed invention with the exception of a compound having a functional group reactive to the curable epoxy compound and a fluorocarbon moiety, wherein the epoxy compound and the compound having a functional group reactive to the curable epoxy compound and a fluorocarbon moiety are polymerized, the fluorocarbon moiety being contained in the resin composition at an amount ranging from 5% to 50% by weight, the compound having a functional group reactive to the curable epoxy compound and a fluorocarbon moiety containing fluorine in an amount of 20% to 80% by weight, the functional group being a hydroxyl group and the compound having a functional group reactive to the curable epoxy compound and a fluorocarbon moiety having a general formula as expressed in claims 6 and 7.

Nakahata et al. teaches (col. 2:7-39) an epoxy resin composition comprising (A) a hydroxyl group containing compound, (B) an epoxy group containing compound, (C) a compound containing a hydrolyzable group directly attached to a silicon atom and /or silanol group and at least one of the above compounds being a fluorine containing resin. Furthermore, the above describes composition also contains a metal chelate as a curing catalyst. Since the metal ions used in the curing process inherently have a positive charge, the polymerization reaction initiated by the metal ions would inherently be a cationic reaction. Nakahata et al. further teaches the compounds in the claimed amounts. The hydroxyl group containing a fluorine moiety, which is analogous to compound having a functional group reactive to the curable epoxy compound and a

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fluorocarbon moiety, is contained in the amount of 20% to 80% (col. 44:22-25).

Furthermore, Table 1 teaches that the hydroxyl containing compound, which is the compound having a functional group reactive to the curable epoxy compound, is given by the formula $\text{CH}_2=\text{CHO}(\text{CH}_2)_4\text{OH}$. This compound has an atomic weight of 116.

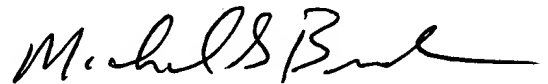
Table 1 also teaches the fluorine moiety is given by the formula $\text{CF}_2=\text{CF}_2$. This compound has an atomic weight of 100. Thus, the compound having a functional group reactive to the curable epoxy compound and a fluorocarbon moiety has a total atomic weight of 216. Fluorine has an atomic weight of 19. Therefore, the total amount of fluorine contained in the compound having a functional group reactive to the curable epoxy compound and a fluorocarbon moiety has an atomic weight of 76. This weight divided by the total weight of 216 gives 35% fluorine in the compound having a functional group reactive to the curable epoxy compound and a fluorocarbon moiety. This epoxy resin composition has the advantages numerous advantages, including improved chemical stability, reduced shrinkage and excellent resistance to environmental conditions (col. 49:3-59 and co. 50:1-5). While this reference does not specifically teach the general formulas of claims 6 and 7, it would have been obvious of design choice to provide a compound having one of these general formulas, since Applicant has not disclosed that a compound having one of these particular formulas solves any stated problem or is for any particular purpose and it appears that the invention would perform equally well with the one of the numerous compounds taught in Nakahata et al.

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3. Any inquiry concerning this communication or earlier communications from the examiner should be directed to Michael S. Brooke whose telephone number is 703-305-0262. The examiner can normally be reached on M-F 5:30-2:00.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, John Barlow can be reached on 308-3126. The fax phone numbers for the organization where this application or proceeding is assigned are 703-308-3431 for regular communications and 703-308-3431 for After Final communications.

Any inquiry of a general nature or relating to the status of this application or proceeding should be directed to the receptionist whose telephone number is 703-305-4900.



Michael S. Brooke
Examiner
Art Unit 2853

MB

MSB
March 28, 2002